Choose all of the following statements that are correct about the variational principle.

- We can only use the variational principle when we explicitly know the stationary state wavefunctions of a quantum system.
- 2) We can always use a trial wavefunction Ψ to estimate an upper bound for the ground state energy of a quantum system.
- 3) We can always use a trial wavefunction Ψ to estimate an upper bound for any excited state energy of a quantum system.

A. 1 only B. 2 only C. 3 only D. 2 and 3 only E. all of the above

Suppose we use the Gaussian function $\Psi(x) = \left(\frac{2b}{\pi}\right)^{1/4} e^{-bx^2}$ (b >

0) as a trial wavefunction to estimate the upper bound for the ground state energy of a delta function potential well $V = -\alpha \delta(x)$. Choose all of the following statements that are correct.

1)
$$\langle \hat{H} \rangle = \langle \hat{T} \rangle + \langle \hat{V} \rangle = \left\langle \Psi \middle| -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} \middle| \Psi \right\rangle + \langle \Psi | V | \Psi \rangle$$

2) $\langle V \rangle = -\alpha \sqrt{\frac{2b}{\pi}}$
3) When $\langle \hat{H} \rangle = \langle \Psi | \hat{H} | \Psi \rangle$ has the minimum value $\langle \hat{H} \rangle_{min}$,
 $\frac{d\langle \hat{H} \rangle_{min}}{db} = 0.$

A. 1 only B. 1 and 2 only C. 1 and 3 only D. 2 and 3 only E. All of the above

Suppose we use the function f(x, b) as the trial wavefunction, where $0 \le b \le 1$ is a variational parameter, to estimate the upper bound for the ground state energy of the Hamiltonian \hat{H} . Choose all of the following statements that are correct.

- 1) The minimum value of $\langle \hat{H} \rangle$ can only be obtained for b = 1 or b = 0.
- 2) The minimum value of $\langle \widehat{H} \rangle$ may be obtained for the value of *b* satisfying $\frac{d\langle \widehat{H} \rangle}{db} = 0.$
- 3) When $\frac{d\langle \hat{H} \rangle}{db} = 0$, $\langle \hat{H} \rangle$ **must** have the minimum value.
- A. 1 only B. 2 only C. 3 only D. 2 and 3 only E. None of the above

The ground state wavefunction of an electron in a hydrogen atom is $\Psi_{100}(r) = \frac{e^{-r/a}}{\sqrt{\pi a^3}}$ (*a* is the Bohr radius) with the ground state energy $E_1 = -13.6 \ eV$. Choose all of the statements that are correct about the two electrons in a **helium** atom.

- 1) The potential energy for the electron-electron interaction is $\frac{e^2}{4\pi\varepsilon_0} \frac{1}{|\vec{r_1} \vec{r_2}|}.$
- 2) The ground state energy of helium is $E_{gs} = 4E_1$ if we ignore the electron-electron interaction.
- 3) The ground state wavefunction $\Psi_0(\vec{r}_1, \vec{r}_2)$ for the two electrons in the helium atom is $\Psi_0(\vec{r}_1, \vec{r}_2) = \Psi_{100}(\vec{r}_1) \cdot \Psi_{100}(\vec{r}_2)$ if we ignore the electron-electron interaction.
- A. 1 only B. 2 only C. 1 and 2 only D. 1 and 3 only E. all of the above

Due to the electron-electron interaction term, the exact solution for the ground-state wavefunction for helium cannot be obtained. We can use a trial wavefunction $\Psi_1(\vec{r}_1, \vec{r}_2) = \frac{Z^3}{\pi a^3} e^{\frac{-Z(r_1+r_2)}{a}}$ to estimate the upper bound energy on the ground state of helium. The parameter Z in the trial function is the effective nuclear charge seen by each electron. Choose all of the following statements that are correct.

1)
$$1 < Z < 2$$

2) $\frac{d\langle \hat{H} \rangle}{dZ} = 0$ when $\langle \hat{H} \rangle = \langle \hat{H} \rangle_{min}$.
3) $\langle \hat{H} \rangle = \langle \hat{T} \rangle + \langle \hat{V} \rangle = \langle \hat{T} \rangle + \langle \Psi | - \frac{e^2}{4\pi\varepsilon_0} \left(\frac{Z}{r_1} + \frac{Z}{r_2} - \frac{1}{|\vec{r_1} - \vec{r_2}|} \right) | \Psi \rangle$

A. 1 only B. 1 and 2 only C. 1 and 3 only D. 2 and 3 only E. all of the above